An Unprecedented Application of Thermally Activated Delayed Fluorescence (TADF) in Light-Emitting Electrochemical Cells (LEECs)

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PHOTOPHYSICAL PROPERTIES



INTRODUCTION

While phosphorescent OLEDs have realized 100% internal quantum efficiency due to recruitment of both singlet and triplet excitons, most of these organometallic emitters are made of expensive metal complexes based on iridium(III) and platinum(II).

Recently, Adachi and co-workers have developed ground-breaking organic fluorescent emitters that operate via a thermality activated delayed fluorescence (TADF) mechanism (*Nature*, **2012**, *492*, 234-238). These emitters harvest both singlet and triplet excitons, which traditional fluorescent emitters are unable to achieve. OLEDs fabricated from these materials have achieved impressive and competitive performance compared to traditional phosphorescent emitters.

Light-emitting electrochemical cells (LEECs) represent another solid-state lighting device architecture that typically employ charged emitters as a key feature. Compared to OLEDs, they enjoy the advantages of using an air-stable cathode, are operated at lower potentials and possess a simpler single-layered device structure.

In this work we illustrated how small molecular charged organic emitters, MYW1 and MYW2, operating through TADF are compatible in LEECs and we also present solution-processed OLED data.

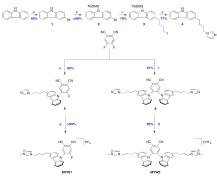


PRIOR ART

In 2012, Adachi and co-workers reported a TADF emitter, **2CzPN**, which achieved an external quantum efficiency (EQE) of 8.0%, breaking the fluorescent emitter EQE limit of 5% (outcoupling efficiency assumed to be 20%), hence suggesting the utilization of triplet excitons. Also shown below is the charged emitter employed in the most recent advance of LEEC devices, featured by its extraordinary long device half-life of 2800 h, recorted by Bolink aroup.



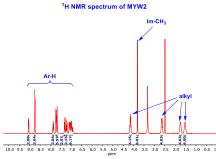


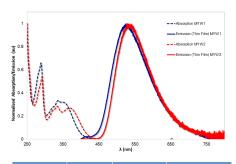


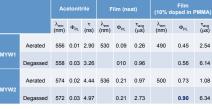
a. NBS, ACN, rt, 1 h. b. i) NaH, THF, rt, 30 min, ii) TBDMSCI c. i) *n*-BuLi, THF, -78 °C, 30 min, ii) excess 1.4-diiodobutane. d. NaH, Imidazole, THF, reflux, 4 h. e. i) NaH, 4, THF, rt, 30 min, ii) 2 equiv. 4,5-diffuorophthalonitrile, rt, 4 h. f. i) NaH, 4, THF, rt, 30 min, ii) 0.6 equiv. 4,5-diffuorophthalonitrile, rt, 4 h. g. i) MeI, ACN, 40 °C, 2h ii) sat. NH₄PF₆ (aq). (NBS = *N*-bromosuccinimide; TBDMSCI = *tert*-butyldimethylsilyl chloride)

STRUCTURAL ELUCIDATION

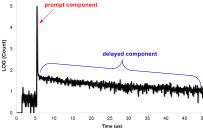
The structures of MYW1 and MYW2 are confirmed by $^1\text{H},~^{13}\text{C}$ and ^{19}F NMR spectroscopy and High-Resolution Mass Spectrometry (HRMS).







MYW2 decay curve (doped film)

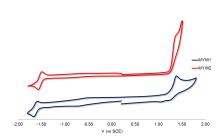


Both MYW1 and MYW2 show TADF characteristics:

 PLQY in N₂ atmosphere always higher than those in air. This is particularly prominent in doped films where ~20% enhancement is observed under a N₂ atmosphere.

Emission decay curves always show both prompt and delayed components.
MYW2 shows a remarkable PLQY of 90% in doped thin film, suggesting its potency in OLED/LECC applications.

ELECTROCHEMICAL PROPERTIES



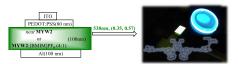
	E ^{ox} _{pa} (V)ª	E ^{red} 1/2 (V)	∆E ^{red} _p (V)	HOMO (eV)	LUMO (eV)	ΔE (V)
MYW1	1.51	-1.50	0.064	-5.93	-2.92	3.01
MYW2	1.45	-1.43	0.079	-5.87	-2.99	2.88

experiments were performed with sample concentration of ca. 10⁺M in 0.1M n-Bu_ANF₂, under M₂-saturated AccN at non temperature with terrocore as the internal standard. The HOMO and LLMO energies were accluded using the relation Energies, and $e^{-i}E_{\rm end} = e^{-i}e$

Both MYW1 and MYW2 demonstrate bipolar characteristics, which are ideal for balanced charge injection and transport.

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LEEC DEVICES



[BMIM]PF6 = 1-butyl-3-methylimida

LEEC Devices ^a	Electroluminescence					
	Avg. Current Density	Lum _{max}	PCEmax	EQE _{max}		
	(A·m-2)	(cd·m·2)a	(Im·W⁻¹) ^b	(%)		
	10	13	0.7	0.39		
1	25	24	0.4	0.29		
	50	26	0.2	0.16		
2	25	10	0.2	0.12		
2	50	17	0.1	0.10		
a Device 1: ITO/PEDOT:PSS/MYW2/AI; Device 2: ITO/PEDOT:PSS/MYW2;[Bmim][PF ₆] 4:1/AI biased with a block wave-pulsed current at a frequency of 1000 Hz and a duty cycle 50% at different current densities						

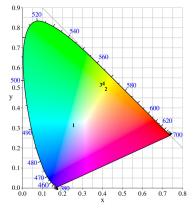
Neat MYW2 performed better than with the addition of ionic liquid (IL):

-Concentration quenching for **MYW2** in LEEC setting is not severe, on the contrary to most OLED emitters which show significant drop in efficiency when applied neat. -Probably, the addition of the polar IL widens $\Delta E_{sr.}$ which decreases TADF efficiency. A very low operating voltage of 2.7 V

OLED DEVICES

			ITO				
		PEDO	DT:PSS(30	nm)			
		Р	VK(30 nm)				
- 1	mCD.		t TADF dy or TADF dye		10)		
	mer.		YMPM(40		10)		
Ca(20 nm)							
Al(100 nm)							
	V	λ	FWHM	EQE,		CE	

OLED Devices	(V)	λ _{peak} (nm)	FWHM (nm)	EQE _{max} (%)	CE _{max} (cd/A)	PE _{max.} (Im/W)	CIE
1 (doped – MYW1)	9.5	467	148	0.3	0.5	0.2	(0.25, 0.31)
2 (doped – MYW2)	5.8	566	137	2.7	7.4	4.0	(0.42, 0.49)
3 (neat – MYW1)	7.5	532	132	0.3	0.7	0.3	(0.38, 0.52)
4 (neat – MYW2)	3.6	546	128	5.1	14.9	10.3	(0.41, 0.53)



Similar to LEEC results, MYW2 outperform MYW1

OLED using neat MYW2 gave consistent, yet surprisingly, better performance than the doped analogue.

A remarkable OLED device was fabricated using *neat* MYW2 as solution-processed emitting layer with high EQE of 5.1% operating at voltage as low as 3.6 V.

CONCLUSIONS

- Two novel TADF-based emitters with charged groups, MYW1 and MYW2, have been synthesized with good overall yields.
- MYW2 was successfully applied in LEEC devices. Preliminary results gave EQE of 0.39% under a very low operating voltage of 2.7 V.
- Solution-processed OLEDs from MeCN using neat MYW2 as the emitting layer gave high EQE of 5.1% with low operating voltage of 3.6 V.
- Surprisingly, in both LEEC and OLED devices, a neat emitting layer invariably outperformed doped analogues, which is a rare observation.
- It is expected that further optimization of LEEC device will give much higher EQE, enjoying both ease of fabrication and high efficiency.

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WLEDGEMENTS AND FUNDING





JM⊠ Johnson Matthey

