Facile and Efficient Chloro-Bridged Iridium (III) Dimers as OLED Materials: Opening U<u>p New</u> Possibilities



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STRUCTURAL ELUCIDATION-C



Iridium chloro-bridged dimers have been long regarded as poor materials for OLED due to their weak emission. Yet, we discovered that we could turn on emission when the dimers were suitably substituted. In this study, a series of four iridium chloro-bridged dimers using formylated ligands have been synthesized and their optoelectronic properties characterized. The OLED device fabricated with **[rt]3-CHO-fppy]**, **CI]**₂ showed a decent external quantum efficiency (EOE) of 2.6%. This study suggests iridium chloro-bridged dimers can be a potential emitter class for OLED apolication.



PRIOR ART

INTRODUCTION

A. M'hamedi, A. S. Batsanov, M. A. Fox, M. R. Bryce, K. Abdullah, H. A. Al-Attarc, A. P. Monkman, J. Mater. Chem., 2012, 22, 13529-13540



SYNTHESIS





r → R 10-B4L eters. 797 10 DMF 10 DMF 10 H⁺ (a) 555 00⁻ 00⁻







11.5 11.0 10.5 10.0 9.5 9.0 8.5 8.0 7.5 7.0 6.5 6.0 ppm





Formyl group on pyridine ring causes a bathochromic shift in the absorption spectra.
 Formyl group placed at the 4-position of the phenyl ring results in a red-shifted absorption spectrum compared with the 3-formyl analogues.

PHOTOPHYSICS PROPERTIES								
Dimer Complex	λ _{em} (nm)	Φ [⊳] (%)	т _е (ns)	k _r (x 10 ⁵ s ⁻¹)	k _{nr} (x 10 ⁵ s ⁻¹)			
[lr(ppy)2Cl]2	520	0.5	125	0.40	79.60			
[Ir(4-CHO-mppy) ₂ CI] ₂	563, 600	2.5	206	1.21	47.33			
[Ir(3-CHO-mppy) ₂ CI] ₂	600	emission too weak						
[lr(4-CHO-fppy)2CI]2	611, 670(sh)	15.7	1980	0.79	4.26			
[lr(3-CHO-fppy)2Cl]2	556	0.9	601	0.15	16.49			
Photophysical measurements were performed in degassed DCM at room temperature.								
$^{\circ}$ Ru(bpv) ₂ (PE ₂) ₂ (Φ_{m} = 9.5%) was used at standard for quantum yield measurements								



 Ir(3-CHO-mppy);Cl];

 Ir(4-CHO-mppy);Cl];

 Ir(hpy);Cl];

 Ir(4-CHO-mppy);Cl];

 Ir(4-CHO-fppy);Cl];

 Ir(4-CHO-fppy);Cl];

 V(vs SCE)

Conditions: complex concentration in 10^{-4} M with 0.1M *n*-Bu₄NPF₈ in N₂-saturated DCM at room temperature; Scan rate: 50 mV s⁻¹.

Complex	E ^{ox} _{pa,1} (V) ^a	E ^{ox} pa,2(V)	E ^{re} _{pc,1} (V)	E ^{re} _{pc,2} (V)	ΔE _{redox} (V)	
[lr(ppy)2Cl]2	1.00	1.26				
[lr(4-CHO-mppy) ₂ Cl] ₂	1.25	1.51				
[lr(3-CHO-mppy) ₂ Cl] ₂	1.35	1.61				
[Ir(4-CHO-fppy) ₂ CI] ₂	1.46	1.67	-1.10	-1.30	2.56	
[lr(3-CHO-fppy) ₂ Cl] ₂	1.58	1.72	-1.16	-1.31	2.74	
* Potentials reported versus SCE.						

OLED FABRICATION

ELECTROCHEMICAL PROPERTIES





Complex	Volt. at 1 cd/m ² (V)	Max. luminance (cd/m²)	Max. Current Eff. (cd/A)	Max. EQE (%)	Emission/peaks (nm)
[lr(3-CHO-fppy) ₂ Cl] ₂	7.7	>6000	9.086	2.593	551
[Ir(3-CHO-mppy)2CI]2	5.7	630	0.508	0.211	557
[Ir(4-CHO-fppy)2CI]2	8.1	178	0.441	0.366	612
[Ir(4-CHO-mppy)2CI]2	6.5	982	1.286	0.637	447, 562

CONCLUSIONS

In this work, the synthesis and characterization of a series of four novel chloro-bridged iridium dimers with formylated ligands were studied. The absorption, electrochemical and photophysical properties were found to be significantly affected by the number of formyl groups and their regiochemistry. More importantly, preliminary OLEDs fabricated using [Ir(3-CHOfppy)₂CIJ₂ as the dopant gave a decent EQE of 2.6%. This result challenges the widely accepted notion that indium dimers cannot be a good OLED materials and encourages further study of chloro-bridged indium dimers as emitters in solid-state lighting.



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